

MG Chemicals UK Limited

Version No: A-2.01

Safety Data Sheet (Conforms to Regulation (EU) No 2015/830)

Issue Date: 08/02/2019 Revision Date: 17/03/2020 L.REACH.GBR.EN

# SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

#### 1.1. Product Identifier

Product name	8329TCF-B	
Synonyms SDS Code: 8329TCF-Part B; 8329TCF-6ML, 8329TCF-50ML, 8329TCF-T50ML, 8329TCF-200ML		
Other means of identification Thermally Conductive Epoxy Adhesive		

#### 1.2. Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Thermally conductive adhesive hardener	
Uses advised against	t Not Applicable	

## 1.3. Details of the supplier of the safety data sheet

Registered company name	MG Chemicals UK Limited	MG Chemicals (Head office)
Address	Hearne House, 23 Bilston Street, Sedgely Dudley DY3 1JA United Kingdom	9347 - 193 Street Surrey V4N 4E7 British Columbia Canada
Telephone	+(44) 1663 362888	+(1) 800-201-8822
Fax	Not Available	+(1) 800-708-9888
Website	Not Available	www.mgchemicals.com
Email	Email sales@mgchemicals.com Info@mgchemicals.com	

#### 1.4. Emergency telephone number

Association / Organisation	Verisk 3E (Access code: 335388)	Not Available
Emergency telephone numbers	+(44) 20 35147487	Not Available
Other emergency telephone numbers	+(0) 800 680 0425	Not Available

#### **SECTION 2 HAZARDS IDENTIFICATION**

2.1.

#### Classification of the substance or mixture

Classification according to regulation (EC) No 1272/2008 [CLP] <sup>[1]</sup>	H314 - Skin Corrosion/Irritation Category 1C, H317 - Skin Sensitizer Category 1B, H412 - Chronic Aquatic Hazard Category 3	
Legend:	1. Classified by Chemwatch; 2. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI	

#### 2.2. Label elements

Hazard pictogram(s)	
SIGNAL WORD	DANGER

#### Hazard statement(s)

H314	Causes severe skin burns and eye damage.	
H317	May cause an allergic skin reaction.	
H412	Harmful to aquatic life with long lasting effects.	

#### Supplementary statement(s)

Not Applicable

P260	Do not breathe dust/fume/gas/mist/vapours/spray.		
P280	Wear protective gloves/protective clothing/eye protection/face protection.		
P273	Avoid release to the environment.		
P272	Contaminated work clothing should not be allowed out of the workplace.		

#### Precautionary statement(s) Response

resolutionary statement(s) response				
P301+P330+P331	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.			
P303+P361+P353	IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].			
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.			
P310	Immediately call a POISON CENTER/doctor/physician/first aider.			
P302+P352	IF ON SKIN: Wash with plenty of water and soap.			
P363	Wash contaminated clothing before reuse.			
P333+P313	If skin irritation or rash occurs: Get medical advice/attention.			
P362+P364	Take off contaminated clothing and wash it before reuse.			
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.			

## Precautionary statement(s) Storage

P405	Store locked up.	

# Precautionary statement(s) Disposal

P501	Dispose of contents/container in accordance with local regulations.
------	---

#### 2.3. Other hazards

REACh - Art.57-59: The mixture does not contain Substances of Very High Concern (SVHC) at the SDS print date.

# SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

#### 3.1.Substances

See 'Composition on ingredients' in Section 3.2

# 3.2.Mixtures

1.CAS No 2.EC No 3.Index No 4.REACH No	%[weight]	Name	Classification according to regulation (EC) No 1272/2008 [CLP]
1.21645-51-2 2.244-492-7 3.Not Available 4.01-2119529246-39-XXXX	57	aluminium hydroxide	Eye Irritation Category 2; H319, EUH066 <sup>[1]</sup>
1.72244-98-5 2.Not Available 3.Not Available 4.Not Available	36	trimercaptan ether, propoxylated	Not Applicable
1.90-72-2 2.202-013-9 3.603-069-00-0 4.01-2119560597-27-XXXX	7	2,4,6- tris[(dimethylamino)methyl]phenol	Acute Toxicity (Oral) Category 4, Skin Corrosion/Irritation Category 2, Eye Irritation Category 2; H302, H315, H319 <sup>[2]</sup>
Legend:	Legend: 1. Classified by Chemwatch; 2. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 3. Classification drawn from C&L * EU IOELVs available		

# SECTION 4 FIRST AID MEASURES

#### 4.1. Description of first aid measures

-	Eye Contact	If this product comes in contact with the eyes: <ul> <li>Immediately hold eyelids apart and flush the eye continuously with running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.</li> <li>Transport to hospital or doctor without delay.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
	Skin Contact	<ul> <li>If skin or hair contact occurs:</li> <li>Immediately flush body and clothes with large amounts of water, using safety shower if available.</li> <li>Quickly remove all contaminated clothing, including footwear.</li> <li>Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.</li> <li>Transport to hospital, or doctor.</li> </ul>
	Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if</li> </ul>

	<ul> <li>necessary.</li> <li>Transport to hospital, or doctor, without delay.</li> <li>Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema.</li> <li>Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs).</li> <li>As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested.</li> <li>Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered.</li> <li>This must definitely be left to a doctor or person authorised by him/her.</li> <li>(ICSC13719)</li> </ul>
Ingestion	<ul> <li>For advice, contact a Poisons Information Centre or a doctor at once.</li> <li>Urgent hospital treatment is likely to be needed.</li> <li>If swallowed do NOT induce vorniting.</li> <li>If voniting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Transport to hospital or doctor without delay.</li> </ul>

#### 4.2 Most important symptoms and effects, both acute and delayed

See Section 11

#### 4.3. Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

For acute or short-term repeated exposures to highly alkaline materials:

- Respiratory stress is uncommon but present occasionally because of soft tissue edema.
- Unless endotracheal intubation can be accomplished under direct vision, cricothyroidotomy or tracheotomy may be necessary.
- Oxygen is given as indicated.
- The presence of shock suggests perforation and mandates an intravenous line and fluid administration.

Damage due to alkaline corrosives occurs by liquefaction necrosis whereby the saponification of fats and solubilisation of proteins allow deep penetration into the tissue.

Alkalis continue to cause damage after exposure.

INGESTION:

# Milk and water are the preferred diluents

No more than 2 glasses of water should be given to an adult.

Neutralising agents should never be given since exothermic heat reaction may compound injury.

\* Catharsis and emesis are absolutely contra-indicated.

\* Activated charcoal does not absorb alkali.

\* Gastric lavage should not be used.

Supportive care involves the following:

Withhold oral feedings initially.

#### If endoscopy confirms transmucosal injury start steroids only within the first 48 hours.

Carefully evaluate the amount of tissue necrosis before assessing the need for surgical intervention.

Patients should be instructed to seek medical attention whenever they develop difficulty in swallowing (dysphagia).

SKIN AND EYE:

Injury should be irrigated for 20-30 minutes.

Eye injuries require saline. [Ellenhorn & Barceloux: Medical Toxicology]

# **SECTION 5 FIREFIGHTING MEASURES**

#### 5.1. Extinguishing media

- Foam.
- Dry chemical powder.
- BCF (where regulations permit).

- Carbon dioxide.
- Water spray or fog Large fires only.

#### 5.2. Special hazards arising from the substrate or mixture

Fire Incompatibility	• Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
5.3. Advice for firefighters	
Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Use fire fighting procedures suitable for surrounding area.</li> <li>Do not approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> <li>Equipment should be thoroughly decontaminated after use.</li> </ul>
Fire/Explosion Hazard	<ul> <li>Combustible solid which burns but propagates flame with difficulty; it is estimated that most organic dusts are combustible (circa 70%) - according to the circumstances under which the combustion process occurs, such materials may cause fires and / or dust explosions.</li> <li>Organic powders when finely divided over a range of concentrations regardless of particulate size or shape and suspended in air or some other oxidizing medium may form explosive dust-air mixtures and result in a fire or dust explosion (including secondary explosions).</li> <li>Avoid generating dust, particularly clouds of dust in a confined or unventilated space as dusts may form an explosive mixture with air, and any source of ignition, i.e. flame or spark, will cause fire or explosion. Dust clouds generated by the fine grinding of the solid are a particular hazard; accumulations of fine dust (420 micron or less) may burn rapidly and fiercely if ignited - particles exceeding this limit will generally not form flammable dust clouds; once initiated, however, larger particles up to 1400 microns diameter will contribute to the propagation of an explosion.</li> <li>In the same way as gases and vapours, dusts in the form of a cloud are only ignitable over a range of concentrations; in principle, the concepts of lower explosive limit (LEL) and upper explosive limit (UEL) are applicable to dust clouds but only the LEL is of practical use; - this is because of the inherent difficulty of achieving homogeneous dust clouds at high temperatures (for dusts the LEL is often called the 'Minimum Explosible Concentration', MEC).</li> <li>When processed with flammable liquids/vapors/mists.ignitable (hybrid) mixtures may be formed with combustible dusts. Ignitable mixtures will increase the rate of explosion pressure rise and the Minimum Ignition Energy (the minimum amount of energy required to ignite dust clouds - MIE) will be lower than the pure dust in air mixture. The Lower Explosive Limit (LEL) of the vapour/dust mixture wi</li></ul>

## Page 4 of 14

#### 8329TCF-B The poxy Adhesive

rmally Conductive E	Ē
---------------------	---

⊢ dusts.
<ul> <li>A dust explosion may release of large quantities of gaseous products; this in turn creates a subsequent pressure rise of explosive force capable of damaging plant and buildings and injuring people.</li> </ul>
Usually the initial or primary explosion takes place in a confined space such as plant or machinery, and can be of sufficient force to damage or rupture the plant. If the shock wave from the primary explosion enters the surrounding area, it will disturb any settled dust layers, forming a second dust cloud, and
often initiate a much larger secondary explosion. All large scale explosions have resulted from chain reactions of this type.
Dry dust can be charged electrostatically by turbulence, pneumatic transport, pouring, in exhaust ducts and during transport.
Build-up of electrostatic charge may be prevented by bonding and grounding.
<ul> <li>Powder handling equipment such as dust collectors, dryers and mills may require additional protection measures such as explosion venting.</li> <li>All movable parts coming in contact with this material should have a speed of less than 1-meter/sec.</li> </ul>
A sudden release of statically charged materials from storage or process equipment, particularly at elevated temperatures and/ or pressure, may result in ignition especially in the absence of an apparent ignition source.
One important effect of the particulate nature of powders is that the surface area and surface structure (and often moisture content) can vary widely from sample to sample, depending of how the powder was manufactured and handled; this means that it is virtually impossible to use flammability data published in the literature for dusts (in contrast to that published for gases and vapours).
<ul> <li>Autoignition temperatures are often quoted for dust clouds (minimum ignition temperature (MIT)) and dust layers (layer ignition temperature (LIT)); LIT generally falls as the thickness of the layer increases.</li> </ul>
Combustion products include:
carbon monoxide (CO)
carbon dioxide (CO2)
nitrogen oxides (NOx)
sulfur oxides (SOx)
other pyrolysis products typical of burning organic material.
May emit corrosive fumes.
Aluminium hydroxide is a flame retardant. At around 200 C, aluminium hydroxide (aluminium trihydrate) is decomposed to aluminium oxide (which forms a
protective, non-flammable layer on the material surface) and water. The water (as steam) forms a layer of non-flammable gas near the material's surface,
inhibiting flames. The reaction is endothermic (absorbs heat energy), thus cooling the material and slowing burning.

# SECTION 6 ACCIDENTAL RELEASE MEASURES

#### 6.1. Personal precautions, protective equipment and emergency procedures See section 8

6.2. Environmental precautions

See section 12

# 6.3. Methods and material for containment and cleaning up

Minor Spills	<ul> <li>Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.</li> <li>Check regularly for spills and leaks.</li> <li>Remove all ignition sources.</li> <li>Clean up all spills immediately.</li> <li>Avoid contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Use dry clean up procedures and avoid generating dust.</li> <li>Place in a suitable, labelled container for waste disposal.</li> </ul>
Major Spills	<ul> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Consider evacuation (or protect in place).</li> <li>Stop leak if safe to do so.</li> <li>Contain spill with sand, earth or vermiculite.</li> <li>Collect recoverable product into labelled containers for recycling.</li> <li>Neutralise/decontaminate residue (see Section 13 for specific agent).</li> <li>Collect solid residues and seal in labelled drums for disposal.</li> <li>Wash area and prevent runoff into drains.</li> <li>After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.</li> <li>If contamination of drains or waterways occurs, advise emergency services.</li> </ul>

#### 6.4. Reference to other sections

Personal Protective Equipment advice is contained in Section 8 of the SDS.

#### SECTION 7 HANDLING AND STORAGE

## 7.1. Precautions for safe handling

Safe handling	<ul> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Avoid contact with moisture.</li> <li>Avoid contact with incompatible materials.</li> <li>When handling, DO NOT eat, drink or smoke.</li> <li>Keep containers securely sealed when not in use.</li> <li>Avoid physical damage to containers.</li> <li>Always wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately. Launder contaminated clothing before re-use.</li> <li>Use good occupational work practice.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.</li> </ul>

	<ul> <li>Organic powders when finely divided over a range of concentrations regardless of particulate size or shape and suspended in air or some other oxidizing medium may form explosive dust-air mixtures and result in a fire or dust explosion (including secondary explosions)</li> <li>Minimise airborne dust and eliminate all ignition sources. Keep away from heat, hot surfaces, sparks, and flame.</li> <li>Establish good housekeeping practices.</li> <li>Remove dust accumulations on a regular basis by vacuuming or gentle sweeping to avoid creating dust clouds.</li> <li>Use continuous suction at points of dust generation to capture and minimise the accumulation of dusts. Particular attention should be given to overhead and hidden horizontal surfaces to minimise the probability of a 'secondary' explosion. According to NFPA Standard 654, dust layers 1/32 in.(0.8 mm) thick can be sufficient to warrant immediate cleaning of the area.</li> <li>Do not use air hoses for cleaning.</li> <li>Minimise dry sweeping to avoid generation of dust clouds. Vacuum dust-accumulating surfaces and remove to a chemical disposal area. Vacuums with explosion-proof motors should be used.</li> <li>Control sources of static electricity. Dusts or their packages may accumulate static charges, and static discharge can be a source of ignition.</li> <li>Solids handling systems must be designed in accordance with applicable standards (e.g. NFPA including 654 and 77) and other national guidance.</li> <li>Do not empty directly into flammable solvents or in the presence of flammable vapors.</li> <li>The operator, the packaging container and all equipment must be grounded with electrical bonding and grounding systems. Plastic bags and plastics cannot be grounded, and antistatic bags do not completely protect against development of static charges.</li> <li>Empty containers may contain residual dust which has the potential to accumulate following settling. Such dusts may explode in the presence of an appropriate ignition source.</li> <li>Do NOT cut, dril</li></ul>
Fire and explosion protection	See section 5
Other information	<ul> <li>Store in original containers.</li> <li>Keep containers securely sealed.</li> <li>Store in a cool, dry, well-ventilated area.</li> <li>Store away from incompatible materials and foodstuff containers.</li> <li>Protect containers against physical damage and check regularly for leaks.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>DO NOT store near acids, or oxidising agents</li> <li>No smoking, naked lights, heat or ignition sources.</li> </ul>

## 7.2. Conditions for safe storage, including any incompatibilities

Suitable container	<ul> <li>Lined metal can, lined metal pail/ can.</li> <li>Plastic pail.</li> <li>Polyliner drum.</li> <li>Packing as recommended by manufacturer.</li> <li>Check all containers are clearly labelled and free from leaks.</li> <li>For low viscosity materials</li> <li>Drums and jericans must be of the non-removable head type.</li> <li>Where a can is to be used as an inner package, the can must have a screwed enclosure.</li> <li>For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):</li> <li>Removable head packaging;</li> <li>Cans with friction closures and</li> <li>low pressure tubes and cartridges</li> <li>may be used.</li> <li>-</li> </ul>
Storage incompatibility	<ul> <li>Avoid strong acids, acid chlorides, acid anhydrides and chloroformates.</li> <li>Avoid contact with copper, aluminium and their alloys.</li> <li>Avoid reaction with oxidising agents</li> </ul>

## 7.3. Specific end use(s)

See section 1.2

## SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

#### 8.1. Control parameters

# DERIVED NO EFFECT LEVEL (DNEL)

Not Available

# PREDICTED NO EFFECT LEVEL (PNEC)

Not Available

# OCCUPATIONAL EXPOSURE LIMITS (OEL)

#### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes	
Not Available	Not Available	Not Available	Not Available	Not Available	Not Available	Not Available	
EMERGENCY LIMITS							
Ingredient	Material name			TEEL-1	TEEL-2	TEEL-3	
aluminium hydroxide	Aluminum hydroxide			8.7 mg/m3	73 mg/m3	440 mg/m3	
2,4,6- tris[(dimethylamino)methyl]phenol	Tris(dimethylaminomethyl)phenol, 2,4,6-			3.6 mg/m3	40 mg/m3	240 mg/m3	
Ingredient	Original IDLH			Revised IDLH			

aluminium hydroxide	Not Available	Not Available	
trimercaptan ether, propoxylated	Not Available	Not Available	
2,4,6- tris[(dimethylamino)methyl]phenol	Not Available	Not Available	

#### MATERIAL DATA

Sensory irritants are chemicals that produce temporary and undesirable side-effects on the eyes, nose or throat. Historically occupational exposure standards for these irritants have been based on observation of workers' responses to various airborne concentrations. Present day expectations require that nearly every individual should be protected against even minor sensory irritation and exposure standards are established using uncertainty factors or safety factors of 5 to 10 or more. On occasion animal no-observable-effect-levels (NOEL) are used to determine these limits where human results are unavailable. An additional approach, typically used by the TLV committee (USA) in determining respiratory standards for this group of chemicals, has been to assign ceiling values (TLV C) to rapidly acting irritants and to assign short-term exposure limits (TLV STELs) when the weight of evidence from irritation, and elimination dure endpoints combine to warrant such a limit. In contrast the MAK Commission (Germany) uses a five-category system based on intensive odour, local irritation, and elimination half-life. However this system is being replaced to be consistent with the European Union (EU) Scientific Committee for Occupational Exposure Limits (SCOEL); this is more closely allied to that of the USA. OSHA (USA) concluded that exposure to sensory irritants can:

- ► cause inflammation
- cause increased susceptibility to other irritants and infectious agents
- lead to permanent injury or dysfunction
- permit greater absorption of hazardous substances and
- + acclimate the worker to the irritant warning properties of these substances thus increasing the risk of overexposure.

It is the goal of the ACGIH (and other Agencies) to recommend TLVs (or their equivalent) for all substances for which there is evidence of health effects at airborne concentrations encountered in the workplace.

At this time no TLV has been established, even though this material may produce adverse health effects (as evidenced in animal experiments or clinical experience). Airborne concentrations must be maintained as low as is practically possible and occupational exposure must be kept to a minimum.

NOTE: The ACGIH occupational exposure standard for Particles Not Otherwise Specified (P.N.O.S) does NOT apply.

#### 8.2. Exposure controls

	<ul> <li>Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:</li> <li>Process controls which involve changing the way a job activity or process is done to reduce the risk.</li> <li>Enclosure and/or isolation of emission source which keeps a selected hazard 'physically' away from the worker and ventilation that strategically 'adds' and 'removes' air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.</li> <li>Employers may need to use multiple types of controls to prevent employee overexposure.</li> <li>Local exhaust ventilation is required where solids are handled as powders or crystals; even when particulates are relatively large, a certain proportion will be powdered by mutual friction.</li> <li>Exhaust ventilation should be designed to prevent accumulation and recirculation of particulates in the workplace.</li> <li>If in spite of local exhaust an adverse concentration of the substance in air could occur, respiratory protection should be considered. Such protection might consist of:</li> <li>(a): particle dust respirators, if necessary, combined with an absorption cartridge;</li> <li>(b): filter respirators with absorption cartridge or canister of the right type;</li> <li>(c): fresh-air hoods or masks</li> <li>Build-up of electrostatic charge on the dust particle, may be prevented by bonding and grounding.</li> <li>Powder handling equipment such as dust collectors, dryers and mills may require additional protection measures such as explosion venting.</li> <li>Air contaminants generated in the workplace possess varying 'escape' velocities which, in turn, determine the 'capture velocities'</li></ul>					
8.2.1. Appropriate engineering controls	Type of Contaminant:		Air Speed:			
	direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas generation into zone of rapid air motion)	s discharge (active	1-2.5 m/s (200-500 f/min.)			
	grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).					
	Within each range the appropriate value depends on:					
	Lower end of the range Upper end of the range					
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents				
	2: Contaminants of low toxicity or of nuisance value only	2: Contaminants of high toxicity				
	3: Intermittent, low production.	3: High production, heavy use				
	4: Large hood or large air mass in motion	4: Small hood-local control only				
	Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 4-10 m/s (800-2000 f/min) for extraction of crusher dusts generated 2 metres distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.					
8.2.2. Personal protection						

Chemical goggles

Eye and face protection

• Full face shield may be required for supplementary but never for primary protection of eyes.

Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands

	▶ thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]
Skin protection	See Hand protection below
Hands/feet protection	<ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gunboots, e.g. Rubber</li> <li>NOTE:</li> <li>The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contract.</li> <li>Contaminated learner items, such as shoes, bells and watch-bands should be removed and destroyed.</li> <li>The selection of subtable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the restance of the glove material can to be calculated in advance and has therefore to be checked prior to the application.</li> <li>The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choics.</li> <li>Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried throughly Application of a non-perfurmed moisturiser is recommended.</li> <li>Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:         <ul> <li>frequency and duration of contact,</li> <li>chemical resistance of glove material,</li> <li>glove thickness and</li> <li>desterity</li> </ul> </li> <li>Select gloves trested to a relevant standard (e.g. Europe EN 374, US F739, ASNZS 2161.1 or national equivalent).</li> <li>When noty brief contact is expected, a glowe with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, ASNZS 2161.1 or national equivalent) is recommended.</li> <li>When noty brief contact is expected, a glowe with a protection class of 3 or higher (breakthrough time sectoring to EN 374, ASNZS 2161.</li></ul>
Body protection	See Other protection below
Other protection	<ul> <li>Overalls.</li> <li>PVC Apron.</li> <li>PVC protective suit may be required if exposure severe.</li> <li>Eyewash unit.</li> <li>Ensure there is ready access to a safety shower.</li> </ul>

#### **Respiratory protection**

Particulate. (AS/NZS 1716 & 1715, EN 143:2000 & 149:001, ANSI Z88 or national equivalent)

Required Minimum Protection Factor up to 10 x ES up to 50 x ES up to 100 x ES	Half-Face Respirator P1 Air-line* Air-line** -	Full-Face Respirator - - P2 P3	Powered Air Respirator PAPR-P1 - PAPR-P2 -
100+ x ES	-	Air-line* Air-line**	- PAPR-P3

\* - Negative pressure demand \*\* - Continuous flow

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

+ Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

- The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option).
- Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.
- Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.

· Use approved positive flow mask if significant quantities of dust becomes airborne.

Try to avoid creating dust conditions.

#### 8.2.3. Environmental exposure controls

See section 12

## SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

## 9.1. Information on basic physical and chemical properties

Appearance	Colorless to pale yellow		
Physical state	Solid	Relative density (Water = 1)	1.59
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	>20.5
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	>93.3	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

#### 9.2. Other information

Not Available

# SECTION 10 STABILITY AND REACTIVITY

10.1.Reactivity	See section 7.2
10.2. Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
10.3. Possibility of hazardous reactions	See section 7.2
10.4. Conditions to avoid	See section 7.2
10.5. Incompatible materials	See section 7.2
10.6. Hazardous decomposition products	See section 5.3

# SECTION 11 TOXICOLOGICAL INFORMATION

## 11.1. Information on toxicological effects

Inhaled	<ul> <li>Evidence shows, or practical experience predicts, that the material produces irritation of the respiratory system, in a substantial number of individuals, following inhalation. In contrast to most organs, the lung is able to respond to a chemical insult by first removing or neutralising the irritant and then repairing the damage. The repair process, which initially evolved to protect mammalian lungs from foreign matter and antigens, may however, produce further lung damage resulting in the impairment of gas exchange, the primary function of the lungs. Respiratory tract irritation often results in an inflammatory response involving the recruitment and activation of many cell types, mainly derived from the vascular system.</li> <li>Inhalation of amine vapours may cause irritation of the mucous membranes of the nose and throat and lung irritation with respiratory distress and cough. Single exposures to near lethal concentrations and repeated exposures to sublethal concentrations produces tracheitis, bronchitis, pneumonitis and pulmonary oedema. Aliphatic and alicyclic amines are generally well absorbed from the respiratory tract. Systemic effects include headache, nausea, faintness and anxiety. These effects are thought to be transient and are probably related to the pharmacodynamic action of the amines. Histamine release by aliphatic amines may produce bronchoconstriction and wheezing.</li> <li>Persons with impaired respiratory function, airway diseases and couditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.</li> <li>If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further risk if handling and use of the material result in excessive exposures.</li> <li>Inhalation hazard is increased at higher temperatures.</li> </ul>
Ingestion	The material can produce chemical burns within the oral cavity and gastrointestinal tract following ingestion. Aliphatic and alicyclic amines are generally well absorbed from the gut. Corrosive action may cause tissue damage throughout the gastrointestinal tract. Detoxification is thought to occur in the liver, kidney and intestinal mucosa with the enzymes, monoamine oxidase and diamine oxidase (histaminase) having a significant role. The material has <b>NOT</b> been classified by EC Directives or other classification systems as 'harmful by ingestion'. This is because of the lack of corroborating animal or human evidence. The material may still be damaging to the health of the individual, following ingestion, especially where pre-existing organ (e.g liver, kidney) damage is evident. Present definitions of harmful or toxic substances are generally based on doses producing mortality rather than those producing morbidity (disease, ill-health). Gastrointestinal tract discomfort may produce nausea and vomiting. In an occupational setting however, ingestion of insignificant quantities is not thought to be cause for concern.

Skin Contact	The material can produce chemical burns following direct contact with the skin. Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions. Volatile amine vapours produce primary skin irritation and dermatitis. Direct local contact, with the lower molecular weight liquids, may produce skin burns. Percutaneous absorption of simple aliphatic amines is known to produce lethal effects often the same as that for oral administration. Cutaneous sensitisation has been recorded chiefly due to ethyleneamines. Histamine release following exposure to many aliphatic amines may result in 'triple response' (white vasoconstriction, red flare and wheal) in human skin. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream through, for example, cuts, abrasions, puncture wounds or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.				
Eye	The material can produce chemical burns to the eye following direct contact. Vapours or mists may be extremely irritating. Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals and/or may produce significant ocular lesions which are present twenty-four hours or more after instillation into the eye(s) of experimental animals. Repeated or prolonged eye contact may cause inflammation characterised by temporary redness (similar to windburn) of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur.				
Chronic	Repeated or prolonged exposure to corrosives may result in the the jaw. Bronchial irritation, with cough, and frequent attacks of exposures may result in dermatitis and/or conjunctivitis. Long-term exposure to respiratory irritants may result in diseas Practical experience shows that skin contact with the material and/or of producing a positive response in experimental anima Limited evidence suggests that repeated or long-term occupati systems. There are reports of lung damage due to excessive inhalation may cause phosphate depletion, especially if phosphate intake softening of the bones. These effects have not been reported in Prolonged or repeated skin contact may cause drying with cra	f bronchial pneumonia may se of the airways involving d is capable either of inducing als. ional exposure may produce of alumina dust. Ingestion of is low. This may cause loss in people occupationally expos	ensue. Gastrointesti lifficult breathing and g a sensitisation read e cumulative health e f large amounts of al of appetite, muscle v sed to aluminium hyo	inal disturbances may also occur. Chronic d related systemic problems. ction in a substantial number of individuals, effects involving organs or biochemical luminium hydroxide for prolonged periods weakness, muscular disease and even droxide.	
8329TCF-B Thermally Conductive		IRRITAT	ION		
Epoxy Adhesive	Not Available	Not Avail	ot Available		
	TOXICITY			IRRITATION	
aluminium hydroxide			Not Available		
trimercaptan ether, propoxylated	ΤΟΧΙΟΙΤΥ	IRRITAT	ION		
	Not Available	Not Avail	able		
	TOXICITY	IDDI	TATION		
	dermal (rat) LD50: >973 mg/kg <sup>[1]</sup>			h - SEVERE	
2,4,6	uermai (rat) LD30. >973 mg/kg* -	Суе (	Eye (rabbit): 0.05 mg/24h - SEVERE		

 2,4,6 IRRITATION

 tris[(dimethylamino)methyl]phenol
 Eye (rabbit): 0.05 mg/24h - SEVERE

 Inhalation (rat) LC50: >0.125 mg/l/1hr.]<sup>[2]</sup>
 Skin (rabbit): 2 mg/24h - SEVERE

 Oral (rat) LD50: 1200 mg/kg<sup>[2]</sup>
 Oral (rat) LD50: 1200 mg/kg<sup>[2]</sup>

data extracted from RTECS - Register of Toxic Effect of chemical Substances

The following information refers to contact allergens as a group and may not be specific to this product. Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other allergic skin reactions, e.g. contact 8329TCF-B Thermally Conductive Epoxy urticaria, involve antibody-mediated immune reactions. The significance of the contact allergen is not simply determined by its sensitisation Adhesive potential: the distribution of the substance and the opportunities for contact with it are equally important. A weakly sensitising substance which is widely distributed can be a more important allergen than one with stronger sensitising potential with which few individuals come into contact. From a clinical point of view, substances are noteworthy if they produce an allergic test reaction in more than 1% of the persons tested. While it is difficult to generalise about the full range of potential health effects posed by exposure to the many different amine compounds, characterised by those used in the manufacture of polyurethane and polyisocyanurate foams, it is agreed that overexposure to the majority of these materials may cause adverse health effects. Many amine-based compounds can induce histamine liberation, which, in turn, can trigger allergic and other physiological effects, including bronchoconstriction or bronchial asthma and rhinitis. > Systemic symptoms include headache, nausea, faintness, anxiety, a decrease in blood pressure, tachycardia (rapid heartbeat), itching, erythema (reddening of the skin), urticaria (hives), and facial edema (swelling). Systemic effects (those affecting the body) that are related to the pharmacological action of amines are usually transient. Typically, there are four routes of possible or potential exposure: inhalation, skin contact, eye contact, and ingestion. Inhalation 2.4.6 Inhalation of vapors may, depending upon the physical and chemical properties of the specific product and the degree and length of exposure, TRIS[(DIMETHYLAMINO)METHYL]PHENOL result in moderate to severe irritation of the tissues of the nose and throat and can irritate the lungs Products with higher vapour pressures have a greater potential for higher airborne concentrations. This increases the probability of worker exposure. Higher concentrations of certain amines can produce severe respiratory irritation, characterised by nasal discharge, coughing, difficulty in breathing, and chest pains. Chronic exposure via inhalation may cause headache, nausea, vomiting, drowsiness, sore throat, bronchopneumonia, and possible lung damage. Also, repeated and/or prolonged exposure to some amines may result in liver disorders, jaundice, and liver enlargement. Some amines have been shown to cause kidney, blood, and central nervous system disorders in laboratory animal studies. While most polyurethane amine catalysts are not sensitisers, some certain individuals may also become sensitized to amines and may experience respiratory distress, including asthma-like attacks, whenever they are subsequently exposed to even very small amounts of vapor.

8329TCF-B Thermally Conduc Adhesi TRIS[(DIMETHYLAMINO)METHYL	ve & 2,4,6-	below hazardous or recommended exposure permanent pulmonary injury, including a reduc Inhalation hazards are increased when expos situations include leaks in fitting or transfer lin bronchitis, and emphysema. <b>Skin Contact:</b> Skin contact with amine catalysts poses a nui simple redness and swelling to painful blister cumulative dermatitis. Skin contact with some amines may result in effects resulting from the absorption of the am blood pressure, reddening of the skin, hives, a amines, and they are usually transient. <b>Eye Contact:</b> Amine catalysts are alkaline in nature and the Direct contact with the liquid amine may caus products may result in mechanical irritation, p Exposed persons may experience excessive The corneal swelling may manifest itself in via a halo phenomenon around lights. These sym Some individuals may experience this effect e <b>Ingestion:</b> The oral toxicity of amine catalysts varies from Some amines can cause severe irritation, uld Material aspirated (due to vorniting) can dam Affected persons also may experience pain in dizziness, drowsiness, thirst, circulatory colla <b>!</b> <b>Polyurethane Amine Catalysts: Guideline Alliance for Polyurethanes Industry</b> The material may produce severe skin irritation to produce conjunctivitis. The material may produce severe skin irritation to since conjunctivitis. The material may produce severe skin irritation to produce conjunctivitis. The material may produce severe skin irritation to produce conjunctivitis. The material may produce severe skin irritation this form of dermatitis is often characterised Histologically there may be intercellular cede contact is unlikely, given the severity of respor Asthma-like symptoms may continue for mon condition known as reactive airways dysfunc compound. Key criteria for the diagnosis of F abrupt onset of persistent asthma-like sympto on spirometry, with the presence of moderate lymphocytic inflammation, without eosinophil irritating inhalation is an infrequent disorder v Industrial bronchitis, on the other hand, is a c	limits should not ordinarily affect healthy i ction in lung function, breathlessness, ch sure to amine catalysts occurs in situation nes. Medical conditions generally aggrav mber of concerns. Direct skin contact cat ing, ulceration, and chemical burns. Rep allergic sensitisation. Sensitised persons nines through skin exposure may include and facial swelling. These symptoms ma sir vapours are irritating to the eyes, even se severe irritation and tissue injury, and pain, and corneal injury.) tearing, burning, conjunctivitis, and com- sual disturbances such as blurred or "fog nptoms are transient and usually disappea aven when exposed to concentrations be n moderately to very toxic. eration, or burns of the mouth, throat, es tage the bronchial tubes and the lungs. In the chest or abdomen, nausea, bleeding pse, coma, and even death. <b>es for Safe Handling and Disposal; Te</b> the eye causing pronounced inflammatic on after prolonged or repeated exposure, by skin redness (erythema) thickening o erma of the spongy layer (spongiosis) and nse, but repeated exposures may produce ths or even years after exposure to the m tion syndrome (RADS) which can occur RADS include the absence of preceding to ms within minutes to hours of a docume to severe bronchial hyperreactivity on mis, have also been included in the criteria with rates related to the concentration of disorder that occurs as result of exposure	nonic bronchitis, and immunologic lung disease. Is that produce aerosols, mists, or heated vapors. Such ated by inhalation exposure include asthma, In cause moderate to severe irritation and injury-i.e., from eated or prolonged exposure may also result in severe Is should avoid all contact with amine catalysts. Systemic headaches, nausea, faintness, anxiety, decrease in y be related to the pharmacological action of the at low concentrations. the "burning" may lead to blindness. (Contact with solid eal swelling. gy" vision with a blue tint ("blue haze") and sometimes ar when exposure ceases. low doses that ordinarily cause respiratory irritation. ophagus, and gastrointestinal tract. g of the throat and the gastrointestinal tract, diarrhea, <b>acchnical Bulletin June 2000</b> on. Repeated or prolonged exposure to irritants may and may produce a contact dermatitis (nonallergic). f the epidermis.
ALUMINIUM HYDROXIDE & TRIMERCAPTAN ETHER, PROPOXYLATED & 2,4,6-		No significant acute toxicological data identified in literature search.		
TRIS[(DIMETHYLAMINO)METHYL	]PHENOL			
Acute Toxicity	×		Carcinogenicity	×
Skin Irritation/Corrosion	<b>~</b>		Reproductivity	×
Serious Eye Damage/Irritation	×		STOT - Single Exposure	×
Respiratory or Skin sensitisation	~		STOT - Repeated Exposure	×
Mutagenicity	×		Aspiration Hazard	×
				ant available on de se vet fill the exitenie for eleccification

Legend: 🗙 – Data

X − Data either not available or does not fill the criteria for classification
→ Data available to make classification

# SECTION 12 ECOLOGICAL INFORMATION

8329TCF-B Thermally Conductive Epoxy Adhesive	ENDPOINT	TEST DURATION (HR)		SPECIES	VAL	UE	SOUR	CE
	Not Available Not Available			Not Available		Not Available		Not Available
aluminium hydroxide	ENDPOINT	TEST DURATION (HR)	SPECIES		VALUE		SOURCE	
	LC50	96	Fish		0.001-0.134mg/L		2	
	EC50	48	Crustacea			0.7364mg/L		2
	EC50	72	Algae or other aquatic plants			0.001-0.05mg/L		2
	NOEC	168	Crustace	a		0.001-mg/L		2
trimercaptan ether, propoxylated	ENDPOINT	TEST DURATION (HR)		SPECIES	VAL	UE	SOUR	CE

Continued...

	Not Available	Not Available		Not Available	Not Available		Not Available
	ENDPOINT	TEST DURATION (HR)	SPE	PECIES		VALUE	SOURCE
2,4,6- tris[(dimethylamino)methyl]phenol	LC50	96	Fish	Fish		175mg/L	2
	EC50	72	Alga	e or other aquatic plants		2.8mg/L	2
		ICLID Toxicity Data 2. Europe ECHA Re	0	0	,		·
		SAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE pan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data					

Harmful to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed of on site or at approved waste sites. Prevent, by any means available, spillage from entering drains or water courses.

**DO NOT** discharge into sewer or waterways.

#### 12.2. Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
2,4,6- tris[(dimethylamino)methyl]phenol	нісн	HIGH

## 12.3. Bioaccumulative potential

Ingredient	Bioaccumulation
2,4,6- tris[(dimethylamino)methyl]phenol	LOW (LogKOW = 0.773)

## 12.4. Mobility in soil

Ingredient	Mobility
2,4,6- tris[(dimethylamino)methyl]phenol	LOW (KOC = 15130)

#### 12.5.Results of PBT and vPvB assessment

	Р	В	т
Relevant available data	Not Applicable	Not Applicable	Not Applicable
PBT Criteria fulfilled?	Not Applicable	Not Applicable	Not Applicable

#### 12.6. Other adverse effects

No data available

#### SECTION 13 DISPOSAL CONSIDERATIONS

#### 13.1. Waste treatment methods

Product / Packaging disposal	<ul> <li>Containers may still present a chemical hazard/ danger when empty.</li> <li>Return to supplier for reuse/ recycling if possible.</li> <li>Otherwise:</li> <li>If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.</li> <li>Where possible retain label warnings and SDS and observe all notices pertaining to the product.</li> <li>DO NOT allow wash water from cleaning or process equipment to enter drains.</li> <li>It may be necessary to collect all wash water for treatment before disposal.</li> <li>In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.</li> <li>Where in doubt contact the responsible authority.</li> <li>Recycle wherever possible.</li> <li>Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.</li> <li>Treat and neutralise at an approved treatment plant.</li> <li>Treatment should involve: Mixing or slurrying in water; Neutralisation with suitable dilute acid followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).</li> <li>Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.</li> </ul>
Waste treatment options	Not Available
Sewage disposal options	Not Available

## **SECTION 14 TRANSPORT INFORMATION**

Labels Required



Limited Quantity: 8329TCF-6ML, 8329TCF-50ML, 8329TCF-200ML

## Land transport (ADR)

14.1. UN number	3259	
4.2. UN proper shipping name	AMINES, SOLID, CORROSIVE, N	I.O.S. or POLYAMINES, SOLID, CORROSIVE, N.O.S. (contains 2,4,6-tris[(dimethylamino)methyl]phenol)
14.3. Transport hazard class(es)	Class 8 Subrisk Not Applicable	
4.4. Packing group	Ш	
4.5. Environmental hazard	Not Applicable	
14.6. Special precautions for user	Hazard identification (Kemler)	80
	Classification code	C8
	Hazard Label	8
	Special provisions	274
	Limited quantity	5 kg

# Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

14.1. UN number	3259			
14.2. UN proper shipping name	AMINES, SOLID, CORR	COSIVE, N.O.S. or POLYAMINES, SOL	D, CORROSIVE,	N.O.S. (contains 2,4,6-tris[(dimethylamino)methyl]phenol)
14.3. Transport hazard class(es)	ICAO/IATA Class ICAO / IATA Subrisk ERG Code	8 Not Applicable 8L		
14.4. Packing group	III			
14.5. Environmental hazard	Not Applicable			
	Special provisions	astructions	A3 A803 864	
	Cargo Only Maximum Qty / Pack		100 kg	
14.6. Special precautions for user	Passenger and Cargo Packing Instructions		860	
	Passenger and Cargo Maximum Qty / Pack		25 kg	
	Passenger and Cargo Limited Quantity Packing Instructions		Y845	
	Passenger and Cargo	Limited Maximum Qty / Pack	5 kg	

## Sea transport (IMDG-Code / GGVSee)

14.1. UN number	3259
14.2. UN proper shipping name	AMINES, SOLID, CORROSIVE, N.O.S. or POLYAMINES, SOLID, CORROSIVE, N.O.S. (contains 2,4,6-tris[(dimethylamino)methyl]phenol)
14.3. Transport hazard class(es)	IMDG Class     8       IMDG Subrisk     Not Applicable
14.4. Packing group	III
14.5. Environmental hazard	Not Applicable
14.6. Special precautions for user	EMS NumberF-A , S-BSpecial provisions223 274Limited Quantities5 kg

# Inland waterways transport (ADN)

14.1. UN number	3259	
14.2. UN proper shipping name	AMINES, SOLID, CORROSIVE, N.O.S. or POLYAMINES, SOLID, CORROSIVE, N.O.S. (contains 2,4,6-tris[(dimethylamino)methyl]phenol)	
14.3. Transport hazard class(es)	8 Not Applicable	
14.4. Packing group	III	
14.5. Environmental hazard	Not Applicable	
14.6. Special precautions for user	Classification code     C8       Special provisions     274	

# 14.7. Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

#### **SECTION 15 REGULATORY INFORMATION**

#### 15.1. Safety, health and environmental regulations / legislation specific for the substance or mixture

#### ALUMINIUM HYDROXIDE(21645-51-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Europe ECHA Registered Substances - Classification and Labelling - DSD-DPD	Europe European Customs Inventory of Chemical Substances ECICS (Romanian)
Europe European Chemicals Agency (ECHA) REACH Registration Numbers	European Chemical Agency (ECHA) Classification & Labelling Inventory - Chemwatch
Europe European Customs Inventory of Chemical Substances - ECICS (Slovak)	Harmonised classification
Europe European Customs Inventory of Chemical Substances ECICS (Bulgarian)	European Customs Inventory of Chemical Substances ECICS (English)
Europe European Customs Inventory of Chemical Substances ECICS (Czech)	European Union - European Inventory of Existing Commercial Chemical Substances (EINECS)
	(English)

#### TRIMERCAPTAN ETHER, PROPOXYLATED(72244-98-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

European Chemical Agency (ECHA) Classification & Labelling Inventory - Chemwatch Harmonised classification

#### 2,4,6-TRIS[(DIMETHYLAMINO)METHYL]PHENOL(90-72-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

ADN - European Agreement concerning the International Carriage of Dangerous Goods by	European Customs Inventory of Chemical Substances ECICS (English)	
Inland Waterways	European Union - European Inventory of Existing Commercial Chemical Substances (EINECS)	
Europe ECHA Registered Substances - Classification and Labelling - DSD-DPD	(English)	
Europe European Agreement concerning the International Carriage of Dangerous Goods by Road - ADR 2017 (Russian)	European Union (EU) Annex I to Directive 67/548/EEC on Classification and Labelling of Dangerous Substances - updated by ATP: 31	
Europe European Chemicals Agency (ECHA) REACH Registration Numbers	European Union (EU) Regulation (EC) No 1272/2008 on Classification, Labelling and	
European Agreement concerning the International Carriage of Dangerous Goods by Road	Packaging of Substances and Mixtures - Annex VI	
(ADR 2011, Norwegian)	European Union (EU) Regulation (EC) No 1272/2008 on Classification, Labelling and Packaging of Substances and Mixtures - Annex VI - Chemwatch Standard Format	
European Agreement concerning the International Carriage of Dangerous Goods by Road		
(ADR 2011, Portuguese)	European Union (EU) Transport of Dangerous Goods by Road - Dangerous Goods List	
European Agreement concerning the International Carriage of Dangerous Goods by Road	(English)	
(ADR 2011, Spanish)	European Union (EU) Transport of Dangerous Goods by Road - Dangerous Goods List	
European Agreement concerning the International Carriage of Dangerous Goods by Road	(French)	
(ADR 2015, German)	European Union (EU) Transport of Dangerous Goods by Road - Dangerous Goods List	
European Agreement concerning the International Carriage of Dangerous Goods by Road	(German)	
(ADR 2017, English)	Regulations concerning the International Carriage of Dangerous Goods by Rail - Table A:	
European Agreement concerning the International Carriage of Dangerous Goods by Road	Dangerous Goods List - RID 2017 (English)	
(ADR 2017, French)	United Nations Recommendations on the Transport of Dangerous Goods Model Regulations	
European Agreement concerning the International Carriage of Dangerous Goods by Road	(Chinese)	
(ADR-S 2019, Swedish)	United Nations Recommendations on the Transport of Dangerous Goods Model Regulations	
European Chemical Agency (ECHA) Classification & Labelling Inventory - Chemwatch	(English)	
Harmonised classification	United Nations Recommendations on the Transport of Dangerous Goods Model Regulations (Spanish)	

This safety data sheet is in compliance with the following EU legislation and its adaptations - as far as applicable - : Directives 98/24/EC, - 92/85/EEC, - 94/33/EC, - 2008/98/EC, - 2010/75/EU; Commission Regulation (EU) 2015/830; Regulation (EC) No 1272/2008 as updated through ATPs.

#### 15.2. Chemical safety assessment

No Chemical Safety Assessment has been carried out for this substance/mixture by the supplier.

#### National Inventory Status

National Inventory	Status
Australia - AICS	Yes
Canada - DSL	Yes
Canada - NDSL	No (2,4,6-tris[(dimethylamino)methyl]phenol; aluminium hydroxide; trimercaptan ether, propoxylated)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	No (trimercaptan ether, propoxylated)
Japan - ENCS	No (trimercaptan ether, propoxylated)
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Legend:	Yes = All ingredients are on the inventory No = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

#### **SECTION 16 OTHER INFORMATION**

Revision Date	17/03/2020
Initial Date	09/02/2019

#### Full text Risk and Hazard codes

H302	Harmful if swallowed.
H315	Causes skin irritation.
H319	Causes serious eye irritation.

#### Other information

#### Ingredients with multiple cas numbers

Name	CAS No
aluminium hydroxide	21645-51-2, 1330-44-5, 1302-29-0, 12252-70-9, 51330-22-4

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chernwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered. For detailed advice on Personal Protective Equipment, refer to the following EU CEN Standards:

EN 166 Personal eye-protection

EN 340 Protective clothing

EN 374 Protective gloves against chemicals and micro-organisms

EN 13832 Footwear protecting against chemicals

EN 133 Respiratory protective devices

#### Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value

LOD: Limit Of Detection

OTV: Odour Threshold Value

**BCF: BioConcentration Factors** 

BEI: Biological Exposure Index

#### **Reason for Change**

A-2.01 - Update to the emergency phone number information.